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Answer: $353 \times 10^5 \text{ kJ h}^{-1}$ (b) If cooling were not provided, the heat of reaction would be absorbed as sensible heat by the streams passing through the reactor. For a rough calculation of the effect of this heat on the temperature of the reactor, assume that the $353 \times 10^5 \text{ kJ h}^{-1}$ is absorbed by 2000 kg h^{-1} ...

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